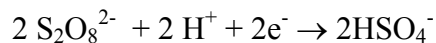


Oxygen BioChem (OBC)TM

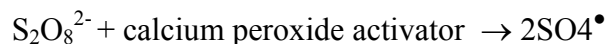
Recent applications of in situ chemical oxidation (ISCO) have shown that ISCO can be a cost-effective remedial strategy for organic contaminants in groundwater and soil. The application of ISCO to contaminated source areas usually results in an immediate benefit to groundwater in the area. In addition, further contaminant flux can be reduced or eliminated mitigating further contaminant plume issues. Redox Tech has recently formulated a mixture of sodium persulfate and calcium peroxide that can be employed for ISCO applications using FMC's patent-pending KlozurTM activation chemistry.

The mixture in OBC supports a two-fold mechanism for treating contaminants of concern. OBC delivers one of the strongest chemical oxidants for short term ISCO, and also provides electron acceptors (oxygen and sulfate) for longer-term biological oxidation.

Sodium persulfate has emerged recently as an important oxidant for in situ remediation of volatile and semi-volatile organic compounds. Persulfate is the strongest oxidant within the peroxygen family, with an electromotive force of 2.12 volts. As illustrated below, the direct oxidation half-cell reaction for persulfate involves a two-electron transfer:

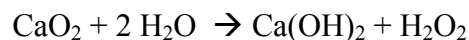


However in most cases, rapid destruction of the contaminant of concern requires that the persulfate be activated in order to generate sulfate radicals. Sulfate radicals are powerful oxidizing agents, with an oxidation potential of 2.6 volts. Klozur activated persulfate is catalyzed with the peroxide and base provide by the calcium peroxide



Activated persulfate can remain available in the subsurface for months providing an unrivaled combination of power and stability.

The calcium peroxide provides several benefits. First, it imparts the alkalinity and peroxide needed to activate the persulfate using FMC's KlozurTM activation chemistry. Second, when mixed with water it provides a long-term slow release source of hydrogen peroxide and calcium hydroxide.



The hydrogen peroxide that is slowly formed decomposes to oxygen and water, providing an extended oxygen source for subsequent bioremediation of petroleum hydrocarbons.

The resultant calcium hydroxide (hydrated lime) that is produced serves several purposes. First of all, it increases the total dissolved ion concentration, which makes the solution less likely to leach metals from the soil into the groundwater. Secondly, the calcium from the hydrated lime will precipitate the sulfate that is produced during the consumption of the

persulfate. The calcium sulfate (gypsum) precipitation helps to reduce sulfate groundwater concentrations, which may impact the secondary drinking water standard of 250 ppm.

The mixture in OBC® provides chemical oxidation as well as electron acceptors (oxygen and sulfate) for longer term biological oxidation. The predominant short-term reaction is chemical oxidation, while the longer-term remediation process is biological oxidation. OBC® has the advantages over more traditional oxygen compounds used for bioremediation in that it works on a broader range of contaminants.

Table 1. Contaminants Treated by Oxygen BioChem (OBC)®

CONTAMINANTS TREATED	
BTEX	1,4-dioxane
MTBE	PCBs
PAHs	Pentachlorophenol
Chlorinated Alkenes	Chlorinated Alkanes

Table 2 summarizes the advantages of OBC® over other oxygenating compounds. OBC® works on a wider range of contaminants than other oxygenates because it provides chemical oxidation and electron acceptors for bioremediation. The amount of oxygen in OBC® assumes that all of the oxygen in the sulfate is consumed through sulfate reduction processes.

Table 2. Advantages of Oxygen BioChem (OBC)®

Oxygen BioChem (OBC)®	COMPETITORS
Greater oxygen – as much as 46 wt %	Typically 10 to 20 wt %
Both chemical oxidation and bioremediation	Predominantly bioremediation
Greater solubility – typically 40 wt % for the persulfate portion	Typically less than 5% soluble
Better value - \$3.25 per pound	Typically \$4 to \$10 per pound

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